

# Take the first step to your success in **IIT-JEE and NEET** with the Pearson Foundation Series **2024**

As per CBSE, ICSE and State curriculum guidelines Includes digital assessment and videos Look inside for more information

Thoroughly revised content

Questions categorised and updated in exercises New feature 'Scientific Crypt' enriches students' engagement



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It offers extensively class-tested content, facilitating result oriented, facilitating efficient preparation to establish a strong foundation for improved academic performance. The pedagogy is designed to encourage learners to go beyond the school curriculum.

# **Salient Features**

Improved learning experience with thoroughly revised theory and exercises

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'Points to Remember' helps review essential concepts

Chapter-end exercises graded based on knowledge, application and analysis



Hints and explanations enhance problem-solving skills



Enriched with online assessments and other digital components



# **IIT Foundation Series Grade VI**



### Science Class 6

- Measurements and Motion
- Force and Pressure
- Work and Energy
- > The Mechanism of Machines
- > Light
- > The Essence of Electricity
- > The Magical Magnetism
- Mystery of Matter

- Language of Chemistry and Changes Around Us
- > Amazing Air
- > Water a wonder liquid
- Environment an eternal gift to mankind
- > Plants nature's wonderful food factories
- Movement an amazing activity
- > Health our greatest wealth

### Mathematics Class 6

- > Numbers
- Factors and Multiples
- > Fractions and Decimals
- Powers and Roots
- Ratio and Proportion
- Percentages and their Applications

- Algebra
- Geometry
- Mensuration
- Statistics
- Sets



# IIT Foundation Series Grade VII

	<ul> <li>Physics Class 7</li> <li>Measurements</li> <li>Kinematics</li> <li>Heat</li> <li>Light</li> </ul>	<ul> <li>Sound</li> <li>Electricity</li> <li>Machines and Tools</li> <li>Our Universe</li> </ul>
Chemistry Cla • Classification of Matte • Atomic Structure and Transformation of Mat • Acids, Bases and Salts	ss 7 • Air and Oxygen • Water • Chemistry in Daily Life	MRP 2 410       Image: Comparison of the second secon
Image: Strategy of the strategy	<ul> <li>Mathematics Class 7</li> <li>Number Systems</li> <li>Expressions and Special Products</li> <li>Ratio and Its Applications</li> <li>Indices</li> <li>Geometry</li> </ul>	<ul> <li>Mensuration</li> <li>Equations and their Applications</li> <li>Formulae</li> <li>Statistics</li> <li>Set Theory</li> </ul>
Biology Cla > Basis of Classification > Nutrition in Plants and > Food and Food Preser > Respiration in Plants a > Transportation in Plant Animals > Reproduction in Plant	<ul> <li>SS 7</li> <li>Our Ecosystem</li> <li>Animals</li> <li>Animal Fibres</li> <li>Soil</li> <li>Weather, Climate and Animal Adaptations</li> <li>Wastewater Management</li> </ul>	Image: State stat

# IIT Foundation Series Grade VIII

	<ul> <li>Physics Class 8</li> <li>Measurements</li> <li>Kinematics</li> <li>Dynamics</li> <li>Hydrostatics</li> <li>Wave Motion and Sound</li> <li>Heat</li> </ul>	<ul> <li>Light</li> <li>Electricity</li> <li>Magnetism</li> <li>Electromagnetism</li> <li>Sources of Energy</li> </ul>
<ul> <li>Chemistry Class 2</li> <li>Atomic Structure</li> <li>Classification of Matter</li> <li>Language of Chemistry and Transformation of Substances</li> <li>Air and Oxygen</li> </ul>	<ul> <li>Water, Solution, Solubility an Hydrogen</li> <li>Carbon and its Compounds</li> <li>Some Important Elements a their Compounds</li> </ul>	and the second s
Mart 60010801 10	Real Numbers and LCM and HCF Squares and Square Roots and Cubes and Cube Roots Indices Polynomials, LCM and HCF of Polynomials Formulae Ratio, Proportion and Variation Percentages Profit and Loss, Discount and Partnership	<ul> <li>Simple Interest and Compound Interest</li> <li>Time and Work, Pipes and Cisterns</li> <li>Time and Distance</li> <li>Linear Equations and Inequations</li> <li>Sets</li> <li>Statistics</li> <li>Geometry</li> <li>Mensuration</li> </ul>
<ul> <li>Biology Class 8</li> <li>Cell- the basic unit of life</li> <li>Tissues and Body Movements</li> <li>Classification of Living Organisms</li> <li>Reproduction in Animals and Plants</li> </ul>	<ul> <li>Microorganisms and Human Health</li> <li>Ecosystem and Biodivers</li> <li>Food- production and management</li> </ul>	sity

# Chapter Insights 🔻





*Konnect* boxes offer add-on information of related topic

# ेर्दू Konnect

A common example of neutralization is our body's response to heartburn.

Antacids are bases that neutralize the excess stomach acid.

# Info Bytes!

The kinetic theory of gases is based on the assumption that gases are composed of a large number of tiny particles (atoms or molecules) that are in constant, random motion. This movement is what gives rise to pressure and temperature. *Info Bytes* are interesting trivia and facts that engage young minds and keep the grey cells ticking!

*Examples* are given in the form of in-text exercises to reinforce the concepts learnt

Illustrative examples solved in a logical and stepwise manner

# SOLUTION

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SOLUTION

SOLUTION Gases differ from solids and liquids in the extent of intermolecular forces of attraction. Since gases are characterized by negligible forces of attraction, the molecules behave independent of the neighboring molecules. Due to large intermolecular spaces, the coefficient of volume expansion is same for all gases. Moreover, in case of gases, the molecules are considered as point masses and hence, the volume occupied by the molecules is negligible in comparison to the total volume. Therefore, the physical behavior of all gases being similar, certain laws could be established which are universally applicable. In case of

liquids and solids, the molecules are considered as rigid spheres and the coefficient of volume expansion is not uniform for all

not he establish

EXAMPLE 1.6 Boyle's law says that pressure and volume are inversely proportional to each other. However, when a balloon is blown, both volume and pressure increase continuously. Justify.

Gas laws are universally applicable for all gases whereas such universal laws could not be established for solids and liquids. Comment on this statement.

hen a balloon is blown continuously, more and more air is forced into the balloon that increasesthe volume as well as the

## (C) POINTS TO REMEMBER

- A symbol is the shorthand notation that represents a single atom of an element.
- A formula is the shorthand notation which represents a single molecule of an element or compound.
- The properties of solids and liquids differ widely from substance to substance, but the gases generally obey some common rules known as the gas laws.
- Boyle's Law states that the volume of a given mass of a gas is inversely proportional to the pressure exerted by the gas at constant temperature.
- Vapor density (VD) of a gas or vapor is the ratio of the mass of a certain volume of gas or vapor to the mass of the same volume of hydrogen gas when their volumes are measured under similar conditions of temperature and pressure.
- A mole is defined as the quantity of substance that contains the same number of elementary particles or chemical units as the number of atoms present in 12 g of C-12 isotope.
- Dalton's law states that 'the total pressure exerted by a mixture of non-reacting gases taken in a container at a given temperature is the sum of the practical pressures of the

*Points to Remember* summarise concepts and topics presented in each chapter



# PRACTICE QUESTIONS CLASSROOM WING

### Very Short Answer Type Questions



Different typologies of questions have been included in the *Classroom Wing* which help students develop problem-solving skills

*Competition Wing* at the end chapters incorporated for competitive examination in-line with JEE syllabus

> Graded as per complexity into Levels 1, 2 and 3.

# **PRACTICE QUESTIONS COMPETITION WING**

### Level 1: Apply your Concepts

Direction for questions from 1 to 8: State whether the following statements are true or false.

1. Doubling the pressure of a gas at constant temperature,

### doubles the volume occupied by the gas. 2. Level 2: Review Your Concepts

- Lever 2. Review Your Concepts
  - 51. Out of oxygen, carbon dioxide, sulphur dioxide ar nitrogen, which among these gases possesses maximu density at a given temperature and pressure?
  - 52. If the Avogadro number is 9  $\times$  10<sup>23</sup>, calculate th weight of 1 mole of CO<sub>2</sub> gas.
  - 53. A gas cylinder is filled with helium at 2000 mm. Du

- 'ous' acid of a non-metal 'X' has 2 'O' atoms. Per acid of the same non-metal has the formula \_\_\_\_\_.
- 17. The ratio of the gram atomic weight of nitrogen and

### oxygen is \_\_\_\_\_. 19 Descure exerted by water vapor in moist gas is called

carree by water vapor in moist gas is cant

### 51. Out of oxygen, carbon dioxide, sulphur dioxide an Level 3: Assess Your Concepts

- 74. At constant temperature and at 4 atm pressure, the given amount of gas occupies 6 L. If the pressure is increased by 6 atm, calculate the percentage change in the volume of the gas.
  75. A gaseous mixture contains 48 g of SO<sub>2</sub> and 33 g of CO<sub>3</sub> at constant temperature and pressure Calculate
- CO<sub>2</sub> at constant temperature and pressure. Calculate the average molecular mass of the gaseous mixture.
   Calculate the ratio of rates of diffusion of three iso-
- topes of hydrogen protium, deuterium and tritium, respectively, at the same temperature and pressure.



# 🗴 🕶 xviii Chapter Insights

# HINTS AND EXPLANATIONS

### Classroom Wing

### Very Short Answer Type Questions

- **1.** 50%, 40%
- 2. higher than
- 3. Relation between vapor density and molecular weight.  $\mathrm{C_2H_6}$
- 4. 16/N
- **5.** 40.043%
- 6. Since the molecular weight of hydrogen is less than



**13.**  $\frac{r_1}{r_2} = \sqrt{\frac{M_2}{M_1}}$ 

12. Mole fraction = No. of moles of a Substance Total number of moles of all substances in the mixture Sum of mole fraction in a mixture = 1 = 0.7 Hints and Explanation for key questions of classroom wing and competition wing along with highlights on the common mistakes that students usually make in the examinations

### Competition Wing

### Level 1: Apply your Concepts

- **20.** Calcium bisulphate  $\rightarrow$  Ca(HSO<sub>4</sub>)<sub>2</sub>
  - Ferric sulphate  $\rightarrow$  Fe<sub>2</sub>(SO<sub>4</sub>)<sub>3</sub>
  - Calcium bisulphite  $\rightarrow$  Ca(HSO<sub>3</sub>)<sub>2</sub>

Ferrous sulphate  $\rightarrow$  FeSO<sub>4</sub>

(i)  $\rightarrow$  (C); (ii)  $\rightarrow$  (D); (iii)  $\rightarrow$  (B); (iv)  $\rightarrow$  (A)

**21.** 80 g of SO is equal to 1 mole. 1 mole of any gas at STP occupies 22.4 L.

Hence, the correct option is (a).

**22.** At low pressure and high temperature, inter-molecular forces of attraction among the gas molecules are negligible. Hence, the gases behave ideally.

Hence, the correct option is (a).

- **23.** (4) HCl + NaOH  $\rightarrow$  NaCl + H<sub>2</sub>O  $\rightarrow$  Double displacement
  - (3)  $Cl_2 + H_2 \rightarrow HCl \rightarrow Combination$

(2) Mg + H<sub>2</sub>SO<sub>4</sub>  $\rightarrow$  MgSO<sub>4</sub> + H<sub>2</sub>  $\rightarrow$  Displacement (1) NaCl Na + Cl  $\rightarrow$  Decomposition (C) potassium bicarbonate  $\rightarrow$  KHCO<sub>3</sub> % of potassium =  $\frac{39}{100} \times 100 = 39\%$ (D) potassium sulphide  $\rightarrow$  K<sub>2</sub>S % of potassium =  $\frac{39 \times 2}{110} \times 100 = 70.90\%$  $\therefore$  increasing order is (D) > (A) > (B) > (C)

Hence, the correct answer is (d).

**38.** Let the vapor density of compound be x Empirical formula weight = x/2

Molecular formula weight = 2x

- • •

 $n = \frac{\text{Molecular formula weight}}{\text{Empirical formula weight}} \times \frac{2x}{\left(\frac{x}{2}\right)} = 4$ 

Molecular formula =  $4 \times A_2B_4 = A_8B_{16}$ Hence, the correct answer is (b).

39. Let the bivalent metal ion be M<sup>2+</sup>, then the formu the compound formed will be M(ClO<sub>4</sub>)<sub>2</sub>.
1 mole of compound contains 8 N oxygen atoms

.

. . . . . .

WORKSHEET

### CHAPTER 1: MOLE CONCEPT, STOICHIOMETRY AND BEHAVIOR OF GASES

### Direction for questions from 1 to 5: Select the correct alternative from the following option.

1. A rubber balloon which is permeable to all the isotopes of hydrogen is filled with pure tritium. It is placed in a closed box containing pure hydrogen under similar conditions of temperature and pressure. Identify the correct observation regarding the balloon.

- (a) The balloon first expands and then contracts.(b) The balloon first contracts first and then
- expand.
- (c) The balloon expands.

### Directions of questions from 6 to 10: State whether the following statements are true or false.

- 6. 80 g of SO3 corresponds to 1 mole.
- 7. Real gases behave ideally at high pressure and low temperature.
- 8. Reaction between an acid and a base form salt and water.
- Equal volumes of all gases possess unequal number of moles at constant temperature and pressure.
- 10. Gas laws are not applicable to the gases at all conditions of temperature and pressure.

*DW: Downloadable Worksheets* supplementing textual exercises for hand-on learning



AI ASSESSMENTS		
AI Assessments	×	
Which of the following is the cheapest natural fiber available in the market?		
O Wool		1
O Silk		
o Jute		AI: Attemptable lests for
		assessing understanding o
		leadiness
		\
		$\sim$
Dusction 1 of 4	Next	

*Video Icon:* Audio and visual learning to difficult questions for effective learning and retention

# Video

Na2CO3 is dissolved in water at 25°C. Which of the following is/are correct regarding the given statement?

(a) In this solution anionic hydrolysis take place.

(b) The concentration of H+ ion in the given solution is less than 10–7 mole/L  $\,$ 

# **IIT Foundation Series Grade IX**

### Physics Class 9

- **> BASIC PHYSICS**
- > Units and Measurements
- **> MECHANICS**
- > Kinematics
- > Dynamics
- Gravitation

MRP ₹420

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IIT FOUNDATION

CHEMISTRY

- > Fluid Mechanics
- **> WAVES AND SOUND**
- > Wave Motion and Sound

### > THERMODYNAMICS

- > Thermodynamics-I
- > ELECTRICITY AND MAGNETISM
- > Electricity-I
- > Magnetism-I
- > LIGHT
- > Light



### **Chemistry** Class 9

- > PHYSICAL CHEMISTRY
- > Nature of Matter
- Basic Concepts of Chemistry
- > Mole Concept, Stoichiometry and Behavior of Gases
- > Atomic Structure
- > Chemical Kinetics and Chemical

- > Equilibrium
- > INORGANIC CHEMISTRY
- Periodic Classification of Elements
- Chemical Bonding
- > Water
- > Metal and Non-Metals

### Mathematics Class 9

- **> NUMBER SYSTEM**
- Real Numbers
- > ARITHMETIC
- > Ratio, Proportion and Variation
- > Percentages, Profit and Loss, **Discount and Partnership**
- Simple Interest and Compound Interest
- > ALGEBRA
- > Polynomial and Square Roots of **Algebraic Expressions**
- Linear Equations in Two Variables
- > Sets and Relations

- Logarithm
- > GEOMETRY
- > Geometry
- Mensuration
- Coordinate Geometry
- > TRIGONOMETRY
- > Trigonometry
- > STATISTICS AND PROBABILITY
- > Statistics
- Probability



# **IIT Foundation Series Grade X**



### Physics Class 10

- **> BASIC PHYSICS**
- > Units and Measurements
- > MECHANICS
- > Dynamics
- Source of Energy
- > WAVES AND SOUND
- > Wave Motion and Sound
- > THERMODYNAMICS
- > Thermodynamics-II

- > ELECTRICITY AND MAGNETISM
- > Electricity-II
- Magnetism-II
- > LIGHT
- > Light
- **> MODERN PHYSICS**
- Modern Physics

MRP **₹595** 

### **Chemistry** Class 10

- > PHYSICAL CHEMISTRY
- > Mole Concept, Stoichiometry and Behavior of Gases
- > Acid, Bases and Salts
- Electrochemistry
- > Periodic Table
- Chemical Bonding

]()



- Industrial Chemistry
- > ORGANIC CHEMISTRY
- > General Organic Chemistry-I
- General Organic Chemistry-II
- > General Organic Chemistry-III



### Mathematics Class 10

- > NUMBER SYSTEM
- > Real Numbers
- > ALGEBRA
- > Polynomials
- > Remainder and Factor Theorems
- Pair of Linear Equations in Two Variables
- > Quadratic Equations
- > Progressions
- > **GEOMETRY**
- > Geometry
- Mensuration
- Coordinate Geometry
- > TRIGONOMETRY
- > Trigonometry

> STATISTICS AND PROBABILITY

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CHEMISTRY

SBN: 9788119847464

- > Statistics
- Probability
- **> COMMERCIAL MATHEMATICS**
- Banking
- Goods and Services Tax (G.S.T.)
- Shares and Dividends
- > BRIDGE TO GRADE XI
- > Statements
- > Sets, Relations and Functions
- > Limits
- Partial Fractions
- > Logarithm

# Introducing Biology Foundation

### **Biology** Class 9

- > Cell- fundamental unit of life
- Tissues
- > Diversity in Living World plants
- > Diversity in Living World animals

### Human Health and Diseases

- Improvement in Food
  - Resources
- Natural Resources

# Image: State of the state



### Biology Class 10

- Cell: The Basic Unit of Life
- · cell. The Busic offic of Elle
- Cell Division and Cell Cycle
- > Life Processes in Plants
- Reproduction in Lower
   Organisms and Plants
- Life Processes in Human Beings-I (digestive and excretory systems)
- > Life Processes in Human

- Beings-II (respiratory and circulatory systems)
- Reproduction in Human Beings
- Control and Coordination
- Genetics
- Evolution
- Ecology
- Management of Natural Resources

# NEET Foundation Series, from Grade VI to X

### **Key Features**

Concepts well-defined with detailed explanations, exhaustive diagrams and images | Cutting-edge info bytes presented in 'Mind-drill' and 'Insight' boxes | Concept-driven subjective and objective-based questions | Practice exercises via Classroom Corner featuring all types of questions |Self-assess your preparation via Competition Corner with around 20% new questions added

### NEET Foundation Class 6, 2023



# Table of Contents

### Physics

Chapter 1: Motion and Measurement Chapter 2: Light, Shadows and Reflections Chapter 3: Electric Current and Circuits Chapter 4: Fun with Magnets

### Chemistry

Chapter 1: Fibre to Fabric Chapter 2: Changes Around Us Chapter 3: Separation of Substances Chapter 4: Sorting Materials into Groups

### Chapter 5: Water

### Biology

Chapter 1: Food and its Sources

Chapter 2: Components of Food

Chapter 3: Fibre to Fabric

Chapter 4: Getting to Know Plants

Chapter 5: Body Movements

**Chapter 6:** The Living Organisms and their Surroundings

Chapter 7: Air Around Us

Chapter 8: Garbage In and Garbage Out

### NEET Foundation Class 7, 2023

### **Physics**

Chapter 1: Flow and Measurement of Heat Chapter 2: Time and Motion Chapter 3: Electric Current and Circuits Chapter 4: Light Chapter 5: Winds, Storms and Cyclones

### Chemistry

Chapter 1: Acids, Bases and Salts Chapter 2: Physical and Chemical Changes Chapter 3: Water: a precious natural resource

### **Table of Contents**

Chapter 4: Wastewater Management

### Biology

Chapter 1: Nutrition in Plants

Chapter 2: Nutrition in Animals

Chapter 3: Animal Fibres

**Chapter 4:** Weather, Climate and Animal Adaptations

Chapter 5: Soil

Chapter 6: Respiration in Organisms

**Chapter 7:** Transportation in Plants and Animals

Chapter 8: Reproduction in Plants

Chapter 9: Forests: our lifeline



### NEET Foundation Class 8, 2023



### Table of Contents

### Physics

Chapter 1: Force and Pressure

Chapter 2: Friction

Chapter 3: Sound

**Chapter 4:** Chemical Effects of Electric Current

Chapter 5: Light Chapter 6: Stars and the Solar System

### Chemistry

Chapter 1: Synthetic Fibres and Plastics
Chapter 2: Metals and Non-metals
Chapter 3: Coal and Petroleum
Chapter 4: Combustion and Flame
Chapter 5: Some Natural Phenomena

### Biology

**Chapter 1:** Crop Production and Management

**Chapter 2:** Microorganisms - friend and foe

**Chapter 3:** Conservation of Plants and Animals

Chapter 4: Cell– structure and functions

Chapter 5: Reproduction in Animals

**Chapter 6:** Reaching the Age of Adolescence

**Chapter 7:** Pollution of Air and Water

### **NEET Foundation Class 9, 2023**

### **Physics**

Chapter 1: Motion Chapter 2: Force and Laws of Motion Chapter 3: Gravitation Chapter 4: Work and Energy Chapter 5: Sound

### Chemistry

Chapter 1: Matter In Our Surroundings Chapter 2: Is Matter Around Us Pure? Chapter 3: Atoms and Molecules Chapter 4: Structure of an Atom

### **Table of Contents**

### Biology

Resources

Chapter 1: The Fundamental Units of Life Chapter 2: Tissues

**Chapter 3:** Diversity in Living Organisms

Chapter 4: Why Do We Fall III?

Chapter 5: Natural Resources Chapter 6: Improvement in Food





### NEET Foundation Class 10, 2023



### Table of Contents

### Physics

**Chapter 1:** Light Reflection and Refraction

**Chapter 2:** The Human Eye and the Colourful World

Chapter 3: Electricity

**Chapter 4:** Magnetic Effects of Electric Current

Chapter 5: Sources of Energy

### Chemistry

**Chapter 1:** Chemical Reactions and Equations

Chapter 2: Acids, Bases and Salts

Chapter 3: Metals and Non-metals

Chapter 4: Carbon and its Compounds

**Chapter 5:** Periodic Classification of Elements

### **Biology**

Chapter 1: Life Processes

Chapter 2: Control and Coordination

**Chapter 3:** How do Organisms Reproduce?

Chapter 4: Heredity and Evolution

Chapter 5: Our Environment

**Chapter 6:** Management of Natural Resources



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